

# A fractal description of pore structure in the SBA family of mesoporous silicates

\*Michael A. Smith and Raul F. Lobo

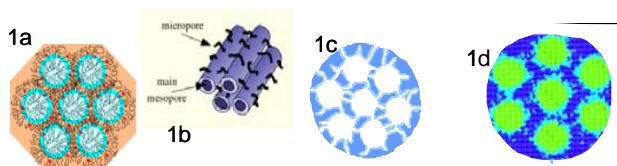
Department of Chemical Engineering, University of Delaware, Newark, DE 19716 and

Department of Chemical Engineering, Villanova University Villanova, PA, 19085

michael.a.smith@villanova.edu

## Introduction

The accurate description of a material's surface properties is necessary to describe molecular processes in catalysis. SBA-15 is a template-synthesized mesoporous silicate that has found extensive use as a model support for studies in supported catalysis. Thorough structural analyses [1,2] clearly describe the dual micropore-mesopore structure with a broad distribution of micropore sizes; however, most artistic renditions of SBA-15 show the micropores as relatively unidimensional, [3-5] as illustrated in Figures 1a, 1b, and 1c. Here we show that the microporosity may alternatively be understood as a manifestation of the fractal nature of the mesopore surface (Figure 1d). A second objective is to utilize the well-defined pore geometry of SBA-15 to compare alternative approaches to calculating fractal dimension from nitrogen adsorption data.



**Figure 1:** Images of SBA-15 from the literature: **1a** from D. Goldfarb, image from [www.bgu.ac.il/.../images/research/sba-15.gif](http://www.bgu.ac.il/.../images/research/sba-15.gif), [4] **1b**, from S. Bhattacharya, image from <http://supriyo.net/research1.htm>, [3]; **1c** from Smith and Lobo[5]. Figure **1d**: fractal recreation of 2D projection of pore surface  $D \sim 2.36$ .

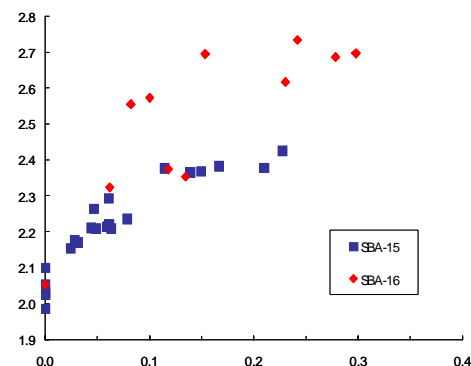
## Methods

SBA-15, SBA-16 and SBA-11 were prepared following a protocol adapted from Zhao et al. [6] We use calcination temperature from 300–1000 °C to adjust the degree of microporosity or fractal character in the SBA materials. Nitrogen sorption isotherms were collected on a Micrometrics ASAP 2010 instrument, and we employ Pfeifer et al's fractal interpretation to multilayer adsorption based upon the Frenkel-Halsey-Hill (FHH) theory for adsorbent - adsorbate interactions in multilayer coverage.

## Results and Discussion

We compare the average fractal dimension calculated using 3 different approaches; only Pfeifer's fractal-FHH approach yields results that are consistent with observations for the range of fractal character in all samples. This fractal dimension obtained from a fractal-FHH

analysis of multilayer adsorption accurately predicts the ratio of BET surface area to the BJH mesopore surface area. This is plotted versus micropore volume in Figure 2 for samples of SBA-15 and SBA-16. SBA-16 shows greater microporosity and a much more significant fractal character, an expected result attributed to the much longer ethylene-oxide fragments in the block co-polymer template. Samples where microporosity has been annealed out by



**Figure 3:** Plot of fractal dimension versus microporosity for SBA-15 and SBA-16 materials

## Significance

The fractal-FHH analysis presented here offers an important perspective on the pore structure of block-copolymer templated mesoporous silicates; moreover, the fractal dimension of the surface provides a quantifiable descriptor of mesopore surface character. We believe this quantifiable descriptor of surface character can provide the basis for understanding support effects in a broad array of supported catalyst studies. Ongoing studies test the hypothesis that catalyst performance should be a function of fractal surface dimension.

## References

1. Kruk, M.; Jaroniec, M.; Ryoo, R.; Kim, J. M., *Chemistry of Materials* 11, 2568 (1999).
2. Imperor-Clerc, M.; Davidson, P.; Davidson, A., *Journal of the American Chemical Society* 122, 11925 (2000).
3. Bhattacharya, S.; Coasne, B.; Hung, F. R.; Gubbins, K. E., *Studies in surface science and catalysis* 527 (2006).
4. Ruthstein, S.; Frydman, V.; Kababya, S.; Landau, M.; Goldfarb, D., *J Phys Chem B* 107, 1739–1748 (2003).
5. Smith, M. A.; Lobo, R. F., *Microporous and Mesoporous Materials* 92, 81 (2006).
6. Zhao, D. Y.; Huo, Q. S.; Feng, J. L.; Chmelka, B. F.; Stucky, G. D., *Journal of the American Chemical Society* 120, 6024 (1998).